

Felix Bergmann, Moshe Weizmann, Elchanan Dimant, Saul Patai, and Jacob Szmuszkovicz. β,β -Diarylacrylic Acids. I. Synthesis and Properties of Symmetrical and Unsymmetrical β,β -Diarylacrylic Acids.

Page 1612. In the line of authors' names and in footnote (1) due to error in the manuscript, Saul Patai appears erroneously as Josef Patai.

Eugene G. Rochow. Methyl Silicate from Silicon and Methanol.

Page 2170. The author writes: "I am indebted to Dr. J. L. Speier for pointing out an error and some inconsistencies in the analytical results given in Table I. The figures shown for 'Product, b. p. 121°' are actually those for a mixture of methanol and methyl silicate, listed with twenty other analyses in the original report and inserted by error into the publication. Typical analyses for the product obtained from the reaction described were H, 8.06; C, 31.4. Calcd. for $\text{Si}(\text{OCH}_3)_4$: H, 7.9; C, 31.6."—E. G. ROCHOW.

James E. LuValle, Dudley B. Glass and Arnold Weissberger. Oxidation Processes. XXI. The Autoxidation of the *p*-Phenylenediamines.

Page 2233. Table III should read:

TABLE III

EFFECT OF SUBSTITUENTS UPON THE RATE OF AUTOXIDATION (RELATIVE RATES)

Compound	pH	pH	$\frac{k_g(pH 11.5)}{k_g(pH 8)}$
	8	11.5	
<i>p</i> -Phenylenediamine	1	29	29
<i>N</i> -Methyl- <i>p</i> -phenylenediamine	25	75 ^a	3
<i>N</i> -Dimethyl- <i>p</i> -phenylenediamine	31	100	3.2
<i>N,N'</i> -Dimethyl- <i>p</i> -phenylenediamine	430	700	1.6
<i>N,N'</i> -Trimethyl- <i>p</i> -phenylenediamine	160	370	2.3
<i>N,N'</i> -Tetramethyl- <i>p</i> -phenylenediamine	110	2	0.018
<i>N</i> -Diethyl- <i>p</i> -phenylenediamine	39 ^b	210	5.4
<i>N</i> -Di- <i>n</i> -propyl- <i>p</i> -phenylenediamine	32
Diaminodurene	410	180	0.44
<i>N</i> -Diethylamino- <i>o</i> -toluidine	210	195	0.93
4-Amino- <i>N,N'</i> -dimethyl- <i>m</i> -toluidine	1	10	10
<i>N</i> -(4-Aminophenyl)-piperidine	1	11	11
<i>N</i> -(4-Aminophenyl)-morpholine	..	3
<i>N</i> -(4-Aminophenyl)-pyrrolidine	28	610	22
<i>N</i> -(4'-Amino-3'-methylphenyl)-piperidine	65	40	0.61
<i>N</i> -(4-Amino-3'-methylphenyl)-morpholine	1	4	4
<i>N</i> -(4'-Amino-3'-methylphenyl)-pyrrolidine	115	1300	11
Octomethyl- <i>p</i> -phenylenediamine	No measurable rate		
^a pH 10.8. ^b pH 7.8.			

H. J. Barber, D. H. O. John and W. R. Wragg. The Reduction of 6-Methoxy-8-(4'-diethylamino-1'-methylbutylidene)-aminoquinoline.

Page 2282. The title was here printed erroneously, with "Methyl-" instead of "Methoxy-."—D. H. O. JOHN.

M. L. Wolfrom and W. J. Polglase. Degradative Studies on Streptomycin.

Page 2835. In col. 2, line 9 from the end, for "N,N,N-tetra-..." read "N,N',N"-tri-..." In line 7 from the end, insert "methyl" following "and."

Page 2836. In col. 1, lines 3 and 5, for "III" read "IV" and for "IV" read "III."—M. L. WOLFROM.

N. Howell Furman and K. G. Stone. A Polarographic Study of Certain Anthraquinones.

Page 3055. In the last line of Table I, the pH value of aqueous borax buffer should read "11" instead of "12."—N. H. FURMAN.

Hugh A. McKenzie. Polarographic Current Time Curves.

Page 3148. In col. 1, line 8, for "0.4" read "4."—H. A. MCKENZIE.

M. L. Wolfrom and C. W. DeWalt. The Configuration of Streptose.

Page 3149. In col. 1, insert at the end of line 13 "H₂O, 12.59." In the next line, after 13.85, insert "H₂O, 12.47."—M. L. WOLFROM.

George Huff, Eleanor Squitieri and Paul E. Snyder. The Heat of Formation of Tungstic Oxide, WO₃.

Page 3381. In Col. 2, the second line of the second equation should read " $\Delta H = -840.1g \pm 0.4$ international kilojoules," and the next equation should read " $\Delta H = -200.84 \pm 0.1g$ kilocalorie." Then the last equation before the Summary should read " $\Delta H_{298.16} = -9.0g \pm 0.1g$ kilocalorie." The same corrections should be made in the summary.—PAUL E. SNYDER.

C. S. Marvel and Jack L. R. Williams. 2-Alkyl-1,3-butadienes.

Page 3843. In formula V, on its left side, the H and CH₂ should be transposed.—C. S. MARVEL.